

Catalytic synthesis of **5**: Imine (0.57 mmol) and acid chloride (0.57 mmol) were combined in 10 mL of CH_3CN and stirred for 15 min. To this solution was added $[\text{Pd}_2(\text{dba})_3] \cdot \text{CHCl}_3$ (5 mol %) in 10 mL of CH_3CN . The reaction mixture was transferred to a 100 mL reaction bomb and left to stir at room temperature for 30 min. 790 Torr of CO was then added to the reaction mixture, and it was allowed to stir at 55°C for 24 h. The resulting solution was filtered through celite, redissolved in CHCl_3 , then washed with dilute HCl, saturated aqueous NaHCO_3 , water, and saturated aqueous NaCl, followed by drying over Na_2SO_4 . After filtration, the solvent was removed in vacuo, and the resultant material dissolved in diethyl ether and cooled to -40°C. The imidazoline **5** was then collected as a white precipitate.

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- [1] F. J. Sardina, H. Rapoport, *Chem. Rev.* **1996**, *96*, 1825.
- [2] a) M. Bodanzsky, A. Bodanzsky, *Practice of Peptide Synthesis*, 2nd ed., Springer, New York, **1994**; b) D. W. Urry, *Angew. Chem.* **1993**, *105*, 859; *Angew. Chem. Int. Ed. Engl.* **1993**, *32*, 819; c) E. Bayer, *Angew. Chem.* **1991**, *103*, 117; *Angew. Chem. Int. Ed. Engl.* **1991**, *30*, 113.
- [3] R. D. Dghaym, K. J. Yaccato, B. A. Arndtsen, *Organometallics* **1998**, *17*, 4.
- [4] Sequential CO/imine insertion with phosphane-coordinated Pd complexes has also been reported: S. Kacker, J. S. Kim, A. Sen, *Angew. Chem.* **1998**, *110*, 1335; *Angew. Chem. Int. Ed.* **1998**, *37*, 1251.
- [5] The Pd-catalyzed coupling of aldehyde, amide, and CO into amino acids has recently been reported: a) M. Beller, M. Eckert, *Angew. Chem.* **2000**, *112*, 1026; *Angew. Chem. Int. Ed.* **2000**, *39*, 1010; b) M. Beller, M. Eckert, F. Vollmller, S. Bogdanovic, H. Geissler, *Angew. Chem.* **1997**, *109*, 1534; *Angew. Chem. Int. Ed. Engl.* **1997**, *36*, 1494.
- [6] a) J. H. Groen, J. G. P. Delis, P. W. N. M. Van Leeuwen, K. Vrieze, *Organometallics* **1997**, *16*, 68; b) R. van Asselt, E. E. C. G. Gielens, R. E. Rulke, K. Vrieze, C. J. Elsevier, *J. Am. Chem. Soc.* **1994**, *116*, 977; c) J. S. Brumbaugh, R. R. Whittle, M. Parvez, A. Sen, *Organometallics* **1990**, *9*, 1735.
- [7] See supporting information for full characterization of **4**, **5**, and **8**. Imidazoline **5** is isolated as a single diastereomer, with a *trans* orientation of the tolyl groups, as determined by X-ray crystallography.
- [8] K. Severin, R. Bergs, W. Beck, *Angew. Chem.* **1998**, *110*, 1722; *Angew. Chem. Int. Ed.* **1998**, *37*, 1634.
- [9] Complex **2** is inert towards either imine deinsertion or scrambling of the imine into the metallacycle.
- [10] a) J. W. Chern, Y. C. Liaw, C. S. Chen, J. G. Rong, C. L. Huang, C. H. Chan, A. H. J. Wang, *Heterocycles* **1993**, *36*, 1091; b) F. Rondu, G. Le Bihan, X. Wang, A. Lamouri, E. Touboul, G. Dive, T. Bellahsene, B. Pfeiffer, P. Renard, B. Guardiola-Lemaitre, D. Manechez, L. Penicaud, A. Ktorza, J. J. Godfroid, *J. Med. Chem.* **1997**, *40*, 3793; c) K. Kameda, S. Ono, Y. Aboko, *Arzneim.-Forsch.* **1982**, *32*, 39; d) J. C. Jonas, M. J. Garcia-Banado, I. Angel, J. C. Henquin, *Eur. J. Pharmacol.* **1994**, *264*, 81; e) I. H. Gilbert, D. C. Rees, A. K. Crockett, R. C. F. Jones, *Tetrahedron*, **1995**, *51*, 6315; f) M. R. Grimmett in *Comprehensive Heterocyclic Chemistry*, Vol. 5 (Eds.: A. Katritzky, C. W. Rees), Pergamon, Oxford, **1984**, p. 457.
- [11] a) H. Han, J. Yoon, K. D. Janda, *J. Org. Chem.* **1998**, *63*, 2045, and references therein; b) P. Merino, A. Lanaspa, F. L. Merchan, T. Tejero, *Tetrahedron Lett.* **1997**, *38*, 1813; c) V. A. Soloshonok, D. V. Avilov, V. P. Kukhar, L. V. Meervelt, N. Mischenko, *Tetrahedron Lett.* **1997**, *38*, 4671.
- [12] a) R. C. F. Jones, A. K. Crockett, D. C. Rees, I. H. Gilbert, *Tetrahedron: Asymmetry* **1994**, *5*, 1661; b) I. H. Gilbert, D. C. Rees, R. S. Richardson, *Tetrahedron Lett.* **1991**, *32*, 2277.
- [13] a) Y. Hsiao, L. S. Hegedus, *J. Org. Chem.* **1997**, *62*, 3586, and references therein; b) C. Betschart, L. S. Hegedus, *J. Am. Chem. Soc.* **1992**, *114*, 5010; c) F. Müller, J. Mattay, *Chem. Ber.* **1993**, *543*; d) *1,3-Dipolar Cycloaddition Chemistry*, Vol. 1 (Ed.: A. Padwa), Wiley, New York, **1984**, p. 177; e) H. Bieräugel, R. Plemp, H. C. Hiemstra, U. K. Pandit, *Tetrahedron* **1983**, *39*, 3971.
- [14] a) T. R. Lin, X. T. Zhou, L. X. Dai, J. Sun, *J. Org. Chem.* **1997**, *62*, 1799; b) X. T. Zhou, Y. R. Lin, L. X. Dai, J. Sun, L. J. Xia, M.-H. Tang, *J.*

Org. Chem. **1999**, *64*, 1331; c) T. Hayashi, E. Kishi, V. A. Soloshonok, Y. Uozumi, *Tetrahedron Lett.* **1996**, *37*, 4969.

- [15] Complex **8** can be independently prepared by reaction of $[\text{Pd}_2(\text{dba})_3] \cdot \text{CHCl}_3$ with **4** under 1 atm ^{13}CO , and is an active catalyst for the reaction.
- [16] a) H. O. Bayer, R. Huisgen, R. Knorr, F. C. Schaefer, *Chem. Ber.* **1970**, *103*, 2581; b) P. D. Croce, R. Ferraccioli, C. La Rosa, *Tetrahedron* **1995**, *51*, 9385; c) P. D. Croce, R. Ferraccioli, C. La Rosa, T. Pilati, *J. Chem. Soc. Perkin Trans. 2* **1993**, *51*, 1511.
- [17] The intermediacy of Münchnone in this chemistry is supported by the independent preparation of **10** ($\text{R}^1 = \text{Bn}$; $\text{R}^2, \text{R}^3 = \text{Ph}$) through dehydration of the appropriate N-protected α -amino acid. Münchnone **10** undergoes a clean reaction with $\text{Tol}(\text{H})\text{C}=\text{NBn}$ to form imidazoline **5** under our acidic reaction conditions.
- [18] For the stoichiometric generation of Münchnone intermediates with iron carbonyl complexes, see: H. Alper, M. Tanaka, *J. Am. Chem. Soc.* **1979**, *101*, 4345.
- [19] The stoichiometric formation of Münchnone intermediates from Fischer carbene complexes has been recently reported: C. A. Merlic, A. Baur, C. C. Aldrich, *J. Am. Chem. Soc.* **2000**, *122*, 7398.
- [20] This process is mechanistically similar to the palladium-catalyzed coupling of aldehydes, amides, and CO into α -amino acids,^[4] although the formation of **5** proceeds under mild conditions, and involves the in situ trapping of the amino acid-based intermediate with imine to form imidazoline.

Synthesis of a Trisaccharide Library by Using a Phenylsulfonate Traceless Linker on Synphase Crowns**

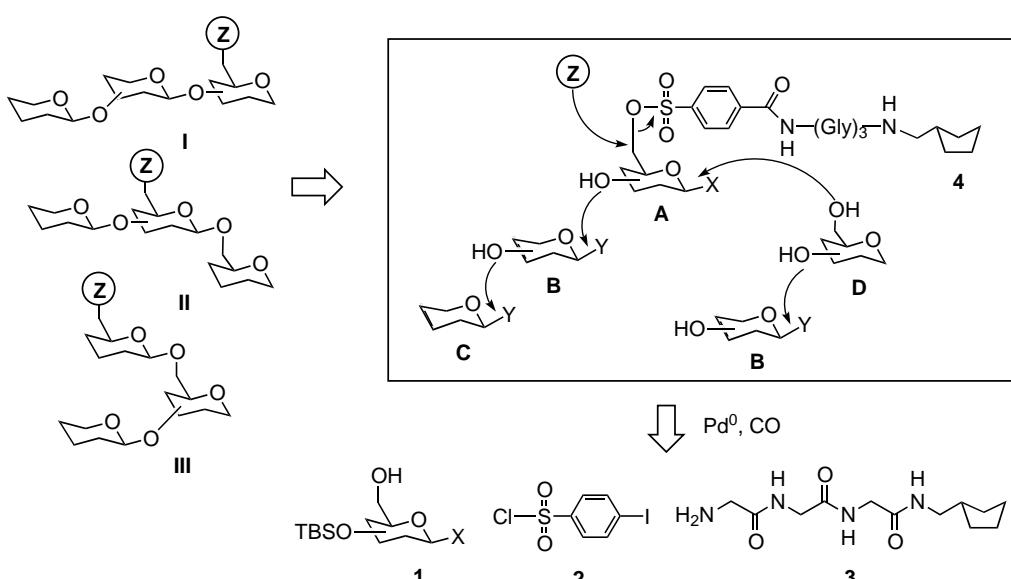
Takashi Takahashi,* Hitoshi Inoue, Yuichi Yamamura, and Takayuki Doi

The development of novel linkers and linkage strategies has become essential in solid-phase synthesis for the discovery of new drugs and materials. In recent years, many efficient linkers were developed.^[1] Traceless linkers are advantageous in that the original functional group of the linker does not remain in the product.^[2] We have reported a phenylsulfonate traceless linker,^[3] which acts as a leaving group under nucleophilic-displacement reaction conditions.^[4,5] With this linker a diversity of products can be obtained, because various functional groups can be introduced at the final stage in a solid-phase synthesis. Herein, we report a high-speed synthesis of a functionalized trisaccharide library utilizing the phenylsulfonate linker on Synphase Crowns.^[6,7]

The synthetic strategy is illustrated in Scheme 1. The trisaccharide derivatives **I**, **II**, and **III** which have various functional groups **Z** at the 6 position of their glucose unit could be synthesized from solid support **4**, which consists of a

[*] Prof. Dr. T. Takahashi, H. Inoue, Y. Yamamura, Prof. Dr. T. Doi
Department of Applied Chemistry
Graduate School of Science and Engineering
Tokyo Institute of Technology
2-12-1 Ookayama, Meguro, Tokyo 152-8552 (Japan)
Fax: (+81) 3-5734-2884
E-mail: ttakashi@o.cc.titech.ac.jp

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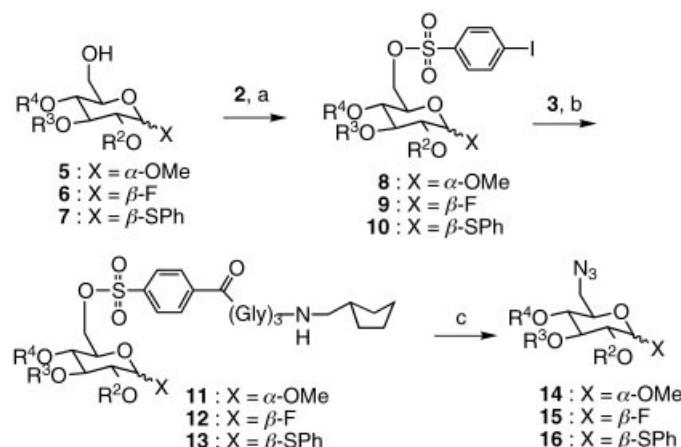
Scheme 1. Synthesis of an oligosaccharide library by using a phenylsulfonate traceless linker. TBS = *tert*-butyldimethylsilyl.

glucose unit **A** linked by a sulfonate linker at the 6 position. 1) Glycosylation of the supported glycosyl acceptor **4** (unit **A**) with glycosyl donor **B** followed by glycosylation of the **B** unit of the produced disaccharide with glycosyl donor **C** and displacement of the sulfonate linker of the **A** unit with a nucleophile **Z** lead to the trisaccharide **I**. 2) Glycosidation of the supported glycosyl donor **4** (unit **A**) with glycosyl acceptor **D** followed by glycosylation of either the **A** or the **D** unit with donor **B** (two-directional glycosylation)^[8] and displacement with **Z** lead to the trisaccharide **II** or **III**. Sulfenylation of monosaccharide **1** at the 6 position with 4-iodobenzenesulfonyl chloride (**2**), which can be regarded as a precursor for an activated ester, was followed by Pd⁰-catalyzed carbonylative amidation with support **3**.^[9] This support, which can be easily handled, consists of aminomethyl crown residues that are distinguishable by the tagging stems clipped to them and a tripeptide spacer (Gly-Gly-Gly) to maintain a suitable distance between the reaction site and the solid support. In this strategy, it is feasible that the diversity of the oligosaccharide library could increase in a combinatorial fashion through variation of the position of the glycosylation and the number of nucleophiles. To synthesize each compound, in the pure form, in the library of trisaccharides, the crown residues were clipped with Trans-tems and were utilized in a split-and-mix method because the crown residues can be identified as and when necessary by “reading” the radiofrequency tags.^[6]

We optimized the reaction conditions of the Pd⁰-catalyzed carbonylative amidation with support **3**.^[10] Methyl glycosides **5**, glycosyl fluorides **6**, and thioglycosides **7** were converted into the precursors **8–10**, respectively, by

sulfenylation with **2** (Scheme 2). The products were carbonylated and amidated with **3**,^[9] the yields were determined by HPLC with integration of the peak area of the corresponding 6-azido-6-deoxymonosaccharide. For methylglycoside **8a** coupling at 40 °C gave, after cleavage, **14a** in 46% yield (Scheme 2, entry 1).^[3, 11] When this carbonylation was carried out at room temperature or at 80 °C (entries 2 and 3, respectively), the yield decreased. The reactions with **8b** and glycosyl fluoride **9e**, both of which have a TBS group at the 3 position, were successful (entries 4 and 5, respectively). The attachment of thioglycoside **10e** onto **3** resulted in decomposition, whereas the attachment of triply benzoyl-substituted thioglycoside **10h** was successful. In the light of these results, we chose to use the glycosyl fluorides **9e–g** as solid-supported glycosyl donors.

A 44-member trisaccharide library was synthesized as follows: carbonylative amidation of methyl glycosides **8b–d** and glycosyl fluorides **9e–g** was carried out in parallel under the conditions given above to afford six different solid-supported monosaccharides, **11b–d** and **12e–g**. The three



Entry	Substrate	T [°C]	Product, yield [%]
1	8a	40	14a , 46
2	8a	RT	14a , 8
3	8a	80	14a , 35
4	8b	40	14b , 48
5	9e	40	15e , 38
6	10e	40	16e , 0
7	10h	40	16h , 35

	R ²	R ³	R ⁴
a	Bn	Bn	Bn
b	Bn	TBS	Bn
c	Bn	Bn	TBS
d	TBS	Bn	Bn
e	Bz	TBS	Bn
f	Bz	Bn	TBS
g	Bz	Bn	Bn
h	Bz	Bz	Bz

Scheme 2. Attachment of glycosides by Pd⁰-catalyzed carbonylative amidation. a) py, CH₂Cl₂; b) ArI (0.5 M), [Pd(PPh₃)₄] (0.01 M), NEt₃ (0.5 M), CO (10 atm), DMF, 40 °C, 24 h; c) NaN₃, DMF, 60 °C, 12 h; Bn = benzyl, Bz = benzoyl, py = pyridine.

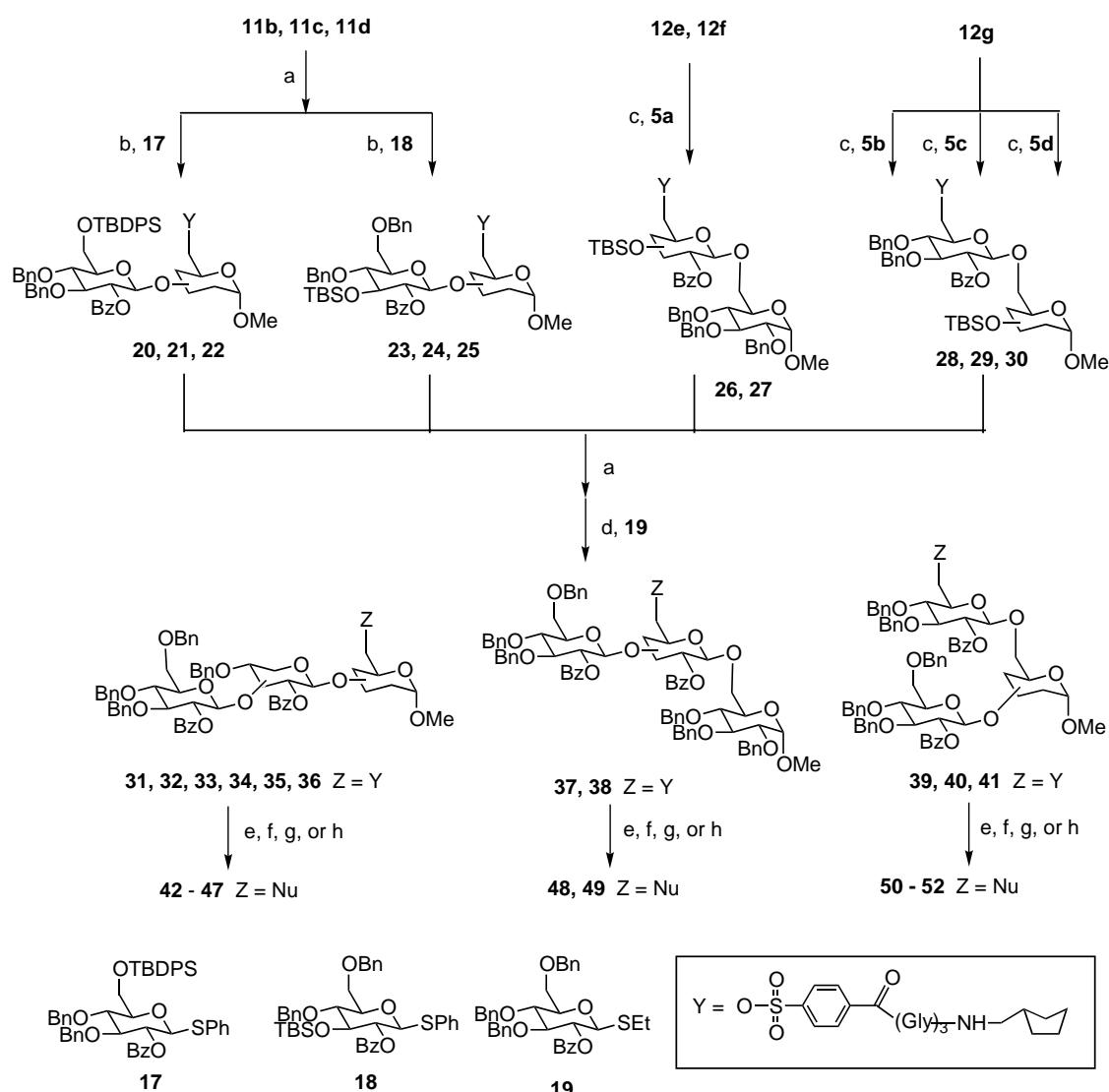
solid-supported methylglycosides **11b–d**, which differ in the position of the TBS group, were mixed, selectively deprotected and divided into two equal fractions (Scheme 3). Using DMTST^[12] as an activator, one fraction was treated with the glycosyl donor **17**, which has a TBDPS group at the 6 position, and the other was treated with the glycosyl donor **18**, which has a TBS group at the 3 position, to provide the six disaccharides **20–25**.^[13] Subsequently, the two solid-supported glycosyl fluorides **12e** and **12f** were placed in a single vial and treated with glycosyl acceptor **5a** and $[\text{Cp}_2\text{Zr}(\text{OTf})_2]$ ^[14] to afford the two disaccharides **26** and **27** (Scheme 3). Next, the solid-supported glycosyl fluoride **12g** was divided into three portions, which were treated with glycosyl acceptor **5b**, **5c**, and **5d**, respectively, in parallel to provide the disaccharides **28–30** (Scheme 3).

The eleven prepared silyl-protected disaccharides **20–30** were combined. Deprotection of the silyl group and glyco-

sylation of all the solid-supported disaccharides with glycosyl donor **19** in a single vial^[15] afforded the solid-supported trisaccharides **31–41**^[16] (Scheme 3).

Finally, the crown compounds were sorted by means of their radiofrequency tags,^[6] and nucleophilic displacement—cleavage with sodium azide, sodium iodide, cesium acetate, and sodium borohydride—furnished the desired 44 functionalized trisaccharides **42–52**.^[3, 17] These reactions were performed in parallel to give each trisaccharide as a pure product, not as a mixture of trisaccharides (Table 1). The purities of the desired trisaccharides were determined by high-performance liquid chromatography (HPLC) and the structures of the respective major products in all the reactions were characterized by mass spectrometry and NMR spectroscopy after preparative HPLC.^[18, 19]

In summary, we have demonstrated a solid-phase synthesis of a functionalized trisaccharide library by using a phenyl-



Scheme 3. Synthesis of a 44-member trisaccharide library by glycosylation of a sulfonate-linked solid support followed by displacement with nucleophiles. a) 2 M HCl in MeOH/THF (1:1), RT; b) glycosyl donor **17** (TBDPS = *tert*-butyldiphenylsilyl) or **18** (0.2 M), DMTST (0.2 M), MS-4 Å, CH_2Cl_2 , RT, 24 h, repeated twice for improving the yield; c) glycosyl acceptor **5** (0.2 M), $[\text{Cp}_2\text{Zr}(\text{OTf})_2]$ (0.2 M), MS-4 Å, CH_2Cl_2 , RT, 24 h; d) glycosyl donor **19** (0.2 M), NIS (0.2 M), TfOH (0.05 M), MS-4 Å, CH_2Cl_2 , RT, 24 h;^[16] e) NaN_3 (0.1 M), DMF, 60 °C, 12 h; f) NaI (0.1 M), DMF, 60 °C, 12 h; g) CsOAc (0.1 M), [18]crown-6 (0.02 M), DMF, 60 °C, 12 h; h) NaBH_4 (0.1 M), DMSO, 60 °C, 12 h. DMTST = dimethyl(methylsulfanyl)sulfonium triflate, NIS = *N*-iodosuccinimide, Tf = trifluoromethanesulfonate.

Table 1. Nucleophilic displacement and cleavage to afford a 44-member trisaccharide library.^[a]

Components	Trisaccharides	Nu = N ₃	Nu = I	Nu = OAc	Nu = H
11b	17 19 Glc β 1 → 6Glc β 1 → 3(6-Nu)Glc α -OMe 42	55	52	49	50
11c	17 19 Glc β 1 → 6Glc β 1 → 4(6-Nu)Glc α -OMe 43	66	70	65	65
11d	17 19 Glc β 1 → 6Glc β 1 → 2(6-Nu)Glc α -OMe 44	51	48	48	46
11b	18 19 Glc β 1 → 3Glc β 1 → 3(6-Nu)Glc α -OMe 45	66	61	60	62
11c	18 19 Glc β 1 → 3Glc β 1 → 4(6-Nu)Glc α -OMe 46	54	52	49	46
11d	18 19 Glc β 1 → 3Glc β 1 → 2(6-Nu)Glc α -OMe 47	71	68	70	65
12e	5a 19 Glc β 1 → 3(6-Nu)Glc β 1 → 6Glc α -OMe 48	83	81	81	75
12f	5a 19 Glc β 1 → 4(6-Nu)Glc β 1 → 6Glc α -OMe 49	50	51	44	43
12g	5c 19 (6-Nu)Glu β 1 → 6Glu α (Glu β 1 → 4)-OMe 50	65	66	63	66
12g	5b 19 (6-Nu)Glu β 1 → 6Glu α (Glu β 1 → 3)-OMe 51	84	79	81	78
12g	5d 19 (6-Nu)Glu β 1 → 6Glu α (Glu β 1 → 2)-OMe 52	92	90	88	90

[a] Purities [%] of the trisaccharides **42**–**52** in the crude products. Glc = D-glucoside.

sulfonate traceless linker. In this study, an efficient strategy for diversification, Pd⁰-catalyzed carbonylative amidation to immobilize monosaccharides, glycosylation at various positions of the solid-supported glycosyl acceptors, and cleavage from the sulfonate linker with four nucleophiles has been developed.

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- [1] I. W. James, *Tetrahedron* **1999**, *55*, 4855–4946; F. Guillier, D. Orain, M. Bradley, *Chem. Rev.* **2000**, *100*, 2091–2157.
- [2] M. J. Plunkett, J. A. Ellman, *J. Org. Chem.* **1995**, *60*, 6006–6007.
- [3] T. Takahashi, S. Tomida, H. Inoue, T. Doi, *Synlett* **1998**, 1261–1263.
- [4] J. A. Hunt, W. R. Roush, *J. Am. Chem. Soc.* **1996**, *118*, 9998–9999.
- [5] J. K. Rueter, S. O. Nortey, E. W. Baxter, G. C. Leo, A. B. Reitz, *Tetrahedron Lett.* **1998**, *39*, 975–978; E. W. Baxter, J. K. Rueter, S. O. Nortey, A. B. Reitz, *Tetrahedron Lett.* **1998**, *39*, 979–982; H. M. Zhong, M. N. Greco, B. E. Maryanoff, *J. Org. Chem.* **1997**, *62*, 9326–9330.
- [6] R. M. Valerio, A. M. Bray, N. J. Maeji, *Int. J. Pept. Protein Res.* **1994**, *44*, 158–165; I. W. James, G. Wickham, N. J. Ede, A. M. Bray in *Solid-Phase Organic Synthesis* (Ed.: K. Burgess), Wiley, New York, **2000**, pp. 195–217.
- [7] The efficient solid-phase glycosylation using an alkylsulfonate linker has been reported. See ref. [4].
- [8] T. Zhu, G.-J. Boons, *Angew. Chem.* **1998**, *110*, 2000–2003; *Angew. Chem. Int. Ed.* **1998**, *37*, 1898–1900.
- [9] T. Takahashi, H. Inoue, S. Tomida, T. Doi, A. M. Bray, *Tetrahedron Lett.* **1999**, *40*, 7843–7846.
- [10] The spacer **3** was prepared from the corresponding aminomethylated crown compound (batch no. 1017-2A, loading = 24.6 μ mol per crown, Mimotopes) by the usual peptide elongation method (Fmoc-glycine, HOEt, DIC; piperidine).
- [11] Displacement of our linker at C-6 of methyl 2,3,4-tri-O-benzyl- α -D-galactoside (NaN₃, DMF, 80 °C, 66 % conversion) gave a 2:1 mixture of the desired azide and an unknown product that was probably produced by the elimination of H-5.
- [12] P. Fügedi, P. J. Garegg, *Carbohydr. Res.* **1986**, *149*, C9–C12.
- [13] All glycosylations on the solid support were carried out as follows: The crown compounds (loading amount of the saccharide 11.3 μ mol per crown) were treated with a CH₂Cl₂ solution of a glycosyl acceptor or donor (20 equiv) pre-dried with MS-4 Å. After 10 min, an activating agent (20 equiv) was added to the mixture as the concentration of both reagents became 0.2 M (1 mL per crown). The mixture was shaken for 24 h at room temperature, then filtered and the residue rinsed with 10 % NEt₃ in CH₂Cl₂. The crown compounds were washed consecutively with CH₂Cl₂, THF, toluene, and CH₂Cl₂. Completion of each reaction step was ascertained by liquid chromatography and electrospray ionization time-of-flight (ESI-TOF) mass spectrometry analysis of the crude product after cleavage from the support (0.1 M NaN₃, DMF, 60 °C, 12 h).

- [14] S. Ikeshita, A. Sakamoto, Y. Nakahara, Y. Nakahara, T. Ogawa, *Tetrahedron Lett.* **1994**, *35*, 3123–3126; T. Matsu-moto, H. Maeta, K. Suzuki, G. Tsuchihashi, *Tetrahedron Lett.* **1988**, *29*, 3567–3578.
- [15] G. H. Veeneman, S. H. van Leeuwen, J. H. van Boom, *Tetrahedron Lett.* **1990**, *31*, 1331–1334.
- [16] Glycosylation of a typical disaccharide produced a trisaccharide with 45 % conversion (with DMTST) and with 83 % conversion (with NIS/TfOH). Two further cycles of the gly-

cosylation under the latter conditions completely consumed the disaccharide on the solid support.

- [17] Sodium borohydride in DMSO reduced **11a** to methyl 2,3,4-O-tribenzyl-6-deoxy- α -D-glucopyranoside in 70 % conversion and 88 % purity; see: J. D. Prugh, A. A. Deana, *Tetrahedron Lett.* **1982**, *23*, 281–284.
- [18] β configuration of the glycosidic linkages in all products was assumed on the basis of the known anchoring effect of an adjacent benzoate group in solid-phase syntheses; selected spectral data for **45** (Nu = OAc): ¹H NMR (CDCl₃, 270 MHz): δ = 1.96 (s, 3 H, acetyl), 3.09 (dd, 1 H, *J* = 3.6, 9.6 Hz, H-2), 3.19 (s, 3 H, methoxy), 3.28 (dd, 1 H, *J* = 7.9, 8.9 Hz, H-4), 3.51–3.86 (m, 10 H), 4.07 (d, 1 H, *J* = 12.2 Hz, benzyl), 4.13–4.58 (m, 15 H), 4.74 (d, 1 H, *J* = 10.6 Hz, benzyl), 4.85 (d, 1 H, *J* = 7.9 Hz, H-1'), 5.02 (d, 1 H, *J* = 11.2 Hz, benzyl), 5.10 (d, 2 H, *J* = 8.9, 8.9 Hz, 2 benzyl), 5.25 (dd, 1 H, *J* = 7.9, 8.9 Hz, H-2'), 5.29 (dd, 1 H, *J* = 7.9, 8.9 Hz, H-2''), 6.96–7.65 (m, 41 H, aromatic), 7.75 (d, 2 H, *J* = 7.3 Hz, aromatic), 7.94 (d, 2 H, *J* = 7.3 Hz, aromatic); ¹³C NMR (CDCl₃, 67.8 MHz): δ = 20.8, 54.9, 63.1, 68.1, 69.1 (2 \times), 73.2, 73.4, 73.5, 74.0, 74.6, 74.8, 74.9, 75.0, 75.1 (3 \times), 75.5, 76.3, 78.1, 78.2, 79.8, 80.9, 83.0, 97.3, 100.5 (2 \times), 127.2, 127.3, 127.4, 127.5, 127.7, 127.8, 127.9, 128.09, 128.14, 128.17, 128.24, 128.3, 128.4, 128.6, 129.7, 129.8, 132.8, 133.2, 137.5, 137.7, 138.0, 138.2, 138.3, 138.4, 138.6, 164.2, 165.2, 170.6; IR (KBr): $\bar{\nu}$ = 3029, 2867, 1732, 1452, 1266, 1070 cm⁻¹; MS (ESI-TOF) calcd for C₈₄H₈₆O₁₉: 1416.6 [M⁺+NH₄], found: 1416.6; **48** (Nu = N₃): ¹H NMR (CDCl₃, 270 MHz): δ = 3.12 (s, 3 H, methoxy), 3.25–3.84 (m, 14 H), 4.03 (br d, 1 H, *J* = 8.9 Hz, H-6), 4.09 (d, 1 H, *J* = 11.2 Hz, benzyl), 4.25 (d, 1 H, *J* = 7.9 Hz, H-1''), 4.29 (d, 1 H, *J* = 10.2 Hz, benzyl), 4.41–5.09 (m, 14 H), 5.06 (d, 1 H, *J* = 11.2 Hz, benzyl), 5.19 (dd, 1 H, *J* = 7.2, 7.9 Hz, H-2''), 5.25 (dd, 1 H, *J* = 8.2, 9.2 Hz, H-2'). 6.90–7.60 (m, 41 H, aromatic), 7.78 (d, 2 H, *J* = 7.8 Hz, aromatic), 7.91 (d, 2 H, *J* = 7.3 Hz, aromatic); ¹³C NMR (CDCl₃, 67.8 MHz): δ = 51.4, 54.8, 67.6, 69.0, 69.1, 73.2, 73.4, 73.8, 74.5, 74.8, 75.1, 75.4, 75.5, 76.3, 77.3, 77.9, 78.8, 79.6, 81.8, 82.8, 97.7, 100.1, 100.5, 127.3, 127.4, 127.5, 127.6, 127.8, 127.9, 128.0, 128.10, 128.14, 128.2, 128.3, 128.4, 128.5, 129.5, 129.6, 129.8, 133.0, 133.2, 137.4, 137.6, 137.8, 138.0, 138.1, 138.2, 138.8, 163.9, 165.4; IR (KBr): $\bar{\nu}$ = 2966, 2812, 2072, 1715, 1439, 1255, 1087 cm⁻¹; MS (ESI-TOF) calcd for C₈₂H₈₃N₃O₁₇: 1399.6 [M⁺+NH₄] found: 1399.7.
- [19] Removal of the protecting groups (benzoyl and benzyl) of two samples **45** (Nu = OAc) and **48** (Nu = N₃) (10 mg cleaved from the three crown compounds; NaOMe, THF/MeOH; Pd(OH)₂/C, MeOH/H₂O) afforded the corresponding free trisaccharides detected by ¹H NMR spectroscopy (400 MHz) and mass spectrometry (ESI-TOF) in quantitative yields. None of the problems mentioned by one of the referees (loss of the product or incomplete hydrogenolysis) was observed.